

Preparation of Bioresorbable Magnesium-substituted Tricalcium Phosphate Ceramics

Xing Zhang and Kenneth S. Vecchio

Materials Science and Engineering Group, University of California, San Diego, La Jolla, CA92093-0411, United States

Introduction:

β -tricalcium phosphate [$\text{Ca}_3(\text{PO}_4)_2$, β -TCP] ceramics have been widely used in dental and skeletal prosthetic applications, due to their good biocompatibility, osteointegration and bioresorbability. However, the poor mechanical properties of β -TCP ceramics, caused by the brittleness and insufficient compaction after sintering below the β - to α - transformation temperature $\sim 1125^\circ\text{C}$, limit the applications. To increase the transition temperature and improve the sintering properties, Mg-substituted tricalcium phosphate [β -TCMP, $(\text{Ca,Mg})_3(\text{PO}_4)_2$] has been produced. With 3 mol% Mg, β -TCMP has the transition temperature above 1300°C . Dense β -TCMP ceramics ($\sim 99\%$ dense) were produced by sintering this β -TCMP green body at 1250°C for 2 hours. In addition, macroporous β -TCMP ceramics were created with sucrose as the porogen.

Methods:

β -TCP powders were prepared by a two-step process. Apatitic tricalcium phosphate [$\text{Ca}_9(\text{HPO}_4)(\text{PO}_4)_5(\text{OH})$] was first prepared through a precipitation method. $\text{Ca}(\text{NO}_3)_2$ solution was slowly titrated into $(\text{NH}_4)_2\text{HPO}_4$ solution (pH ~ 10) at room temperature. The precipitates were aged for 1 hour, harvested and dried at 80°C for 12 hours. This β -TCP 'precursor' was then heated at 800°C for 3 hours to produce β -TCP. To prepare β -TCMP, varying amounts of $\text{Mg}(\text{NO}_3)_2 \cdot 6\text{H}_2\text{O}$ were added into $\text{Ca}(\text{NO}_3)_2$ solution as the Mg source. For dense ceramics, β -TCMP powder was grinded with addition of drops of 5wt.% polyvinyl alcohol (PVA) solution. The powder was then uniaxially pressed in a stainless steel die on a load frame at different pressures. The green bodies were further sintered in air at 1250°C for 2 hours. For macroporous β -TCMP ceramics, sucrose was mixed with β -TCMP powder as the porogen to make a green body. The green body was then slowly heated up to burn off sucrose and further sintered at 1250°C to improve structural integrity of the porous body. The samples were characterized by XRD, TG/DTA, SEM, EDS and FT-IR techniques. The relative density of the compacted ceramics was measured following Archimedes' method. Quasi-static compression tests were conducted at a strain rate of $10^{-3}/\text{s}$.

Results:

β -TCMP precursor was prepared by the precipitation method at room temperature (Fig. 1a). Pure β -TCMP can be prepared by heating this precursor at 800°C for 3 hours (Fig. 1b). With increase of the amount of Mg, the β - to α -transition temperature increases. β -TCMP with 3mol%Mg has a transition temperature above 1300°C (Fig. 1c).

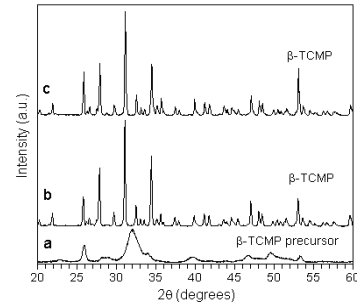


Fig. 1. XRD pattern of (a) β -TCMP (3mol%Mg) precursor, (b) β -TCMP produced by heating the precursor at 800°C for 3 hours and (c) heated at 1300°C for 2 hours.

The ceramic density is affected by the loading pressure for the green body. With a loading pressure of 60MPa and 150MPa, the relative density of β -TCMP ceramics after sintered at 1250°C are $\sim 93\%$ and 99% , respectively (Table 1). The average compressive strength of $\sim 99\%$ relative dense β -TCMP ceramics is $\sim 190\text{MPa}$.

Table 1. The effect of loading pressure to the density of β -TCMP ceramics (3mol%Mg) sintered at 1250°C for 2h.

Sample	Loading pressure	Relative density
1	60MPa	93%
2	100MPa	97%
3	150MPa	99%

Macroporous β -TCMP ceramics (pore size $\sim 100\text{-}300\mu\text{m}$) can be produced from β -TCMP or 'precursor' powder with sucrose as the porogen (Fig. 2). The ceramics were heated up at $2^\circ\text{C}/\text{min}$ and sintered at 1250°C for 2h in air.

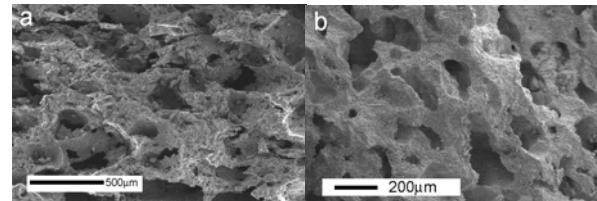


Fig. 2. SEM images of macroporous β -TCMP ceramics created from (a) β -TCMP (3 mol% Mg) precursor, (b) β -TCMP (5 mol% Mg) powder with sucrose as the porogen.

Conclusions:

β -TCMP was produced by a two-step process: precipitation and further heating of β -TCMP precursor. With the increase of the amount of Mg, β - to α - transition temperature increases. β -TCMP with 3mol%Mg has a transition temperature above 1300°C . 99% relative dense β -TCMP (3mol%Mg) can be produced with a compaction pressure of 150MPa and further sintering at 1250°C for 2 hours. The average compressive strength of 99% dense β -TCMP is $\sim 190\text{MPa}$. Macroporous β -TCMP ceramics can be created with sucrose as a porogen.